

Rotational Isomers of a 1,2-Diphenyl-3,4-diphosphinidenecyclobutene

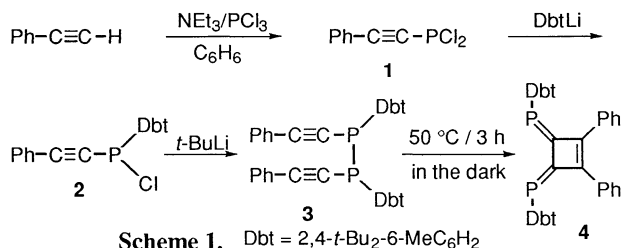
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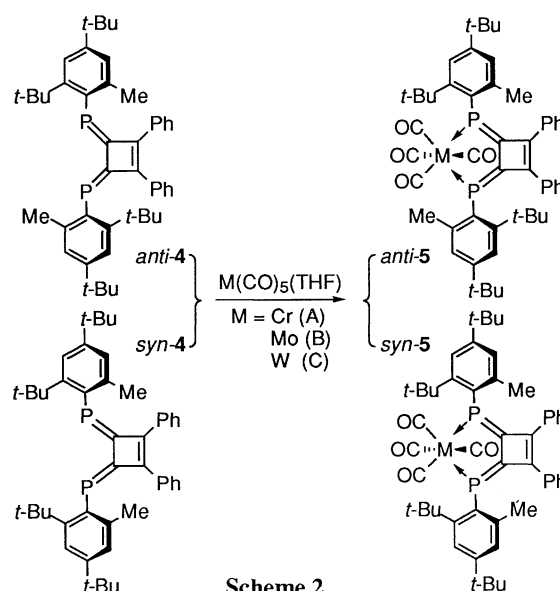
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The group-6 metal tetracarbonyl complexes of 3,4-diphosphinidenecyclobutenes protected with the 2,4-di-*t*-butyl-6-methylphenyl group were prepared and one of the rotamers, ligated with W(CO)₄, was confirmed by X-ray analysis while the other was analyzed by a chiral HPLC column.

Using an extremely bulky 2,4,6-tri-*t*-butylphenyl group (hereafter, abbreviated to the Ar group), we were successful in preparation of several unusual organophosphorus compounds¹ such as diphosphenes,^{2a} phosphalkenes,^{2b} and phosphacumulenes^{2c,d} as stable species. Appel,³ Märkl,⁴ and we⁵ prepared some sterically protected diphosphinidenecyclobutenes as an additional example of low coordinated phosphorus compounds, which belongs to a phosphorus analogue of dimethylenecyclobutene. Furthermore, we have reported on the preparation and structures of the group-6 metal(0)⁶ and palladium(II)⁷ complexes of diphosphinidenecyclobutenes ligating as a bidentate ligand. On the other hand, we have developed several protecting groups other than the Ar group and found that the 2,4-di-*t*-butyl-6-methylphenyl (hereafter abbreviated to Dbt) group is useful to stabilize some unsymmetrical diphosphene⁸ and phosphathenes.^{9,10} We made a further application of the Dbt group to the diphosphinidenecyclobutene system and found conformational isomerism due to the restricted rotation around the two P-Dbt bonds at the edges of the system.



The diphosphinidenecyclobutene was prepared as described in Scheme 1. Phenylacetylene was converted to phenylethynylphosphonous dichloride **1** ($\delta_P = 122$)¹¹ in 24% and **1** was allowed to react with DbtLi to give the corresponding phosphinoyl chloride **2** ($\delta_P = 46$). The chloride **2** was then allowed to react with equimolar amount of *t*-butyllithium to give 1,2-bis(2,4-di-*t*-butyl-6-methylphenyl)-1,2-bis(phenylethynyl)diphosphane **3** ($\delta_P = -53$). When the diphosphane **3** was heated in the dark for 3 h, 3,4-bis(2,4-di-*t*-butyl-6-methylphenyl)-1,2-diphenylcyclobutene **4** was obtained. It is interesting to note that the compound **4** consists of two rotamers, *syn*-**4** and *anti*-**4**, the ³¹P NMR of which appeared as two singlets, δ_P at 169.1 and 168.4 in a ratio of 1 : 2, but attempts to separate these rotamers were unsuccessful.



After addition of M(CO)₅(THF) to this mixture, where M equals Cr (A), Mo (B), and W (C), the ³¹P NMR of the reaction products appeared as two singlets again in a ratio of about 1 : 2, respectively, due to *syn*-**5** and *anti*-**5** with complete retention of conformation (Scheme 2). The compounds **5A–C** were purified through column chromatography to give dark purple complexes and were analyzed by spectroscopic data.¹²

Furthermore, the structure of *syn*-**5C** was unambiguously determined by X-ray analysis.^{13,14} Figure 1 depicts a molecular structure of *syn*-**5C**, which clearly shows the ligand on the tungsten metal is in the *syn*-configuration. The atoms W, P(1), C(1), C(2), and P(2) are on the same plane with a mean deviation of 0.01 Å and the atoms C(1), C(2), C(3), and C(4) are coplanar with a mean deviation of 0.01 Å making a dihedral angle of 2.9°, each other. Thus, the diphosphinidenecyclobutene framework including the W atom is almost coplanar. The two Dbt rings (C(5)–C(10) and C(20)–C(25)) are almost perpendicular to the WPCCP plane making interplanar angles of 100° and 94°, respectively. The two phenyl rings, C(35)–C(40) and C(41)–C(46), make dihedral angles of 37.2° and 18.7°, with the cyclobutene ring, respectively. The six ligands around the W atom are located forming a slightly distorted octahedron. The P=C bond lengths (1.669(6) Å for P(1)–C(1) and 1.662(6) Å for P(2)–C(2), respectively) are close to the values for the corresponding complexes as reported for 3,4-bis(2,4,6-tri-*t*-butylphenylphosphinidene)-1,2-bis(trimethylsilyl)cyclobutene (**6**) (1.678(6) and 1.676(5) Å),^{6b} and the bond length C(1)–C(2) for *syn*-**5C** is 1.507(8) Å, whereas that for **6**·[Mo(CO)₄] is 1.483(6) Å.^{6b} The X-ray analysis indicated that

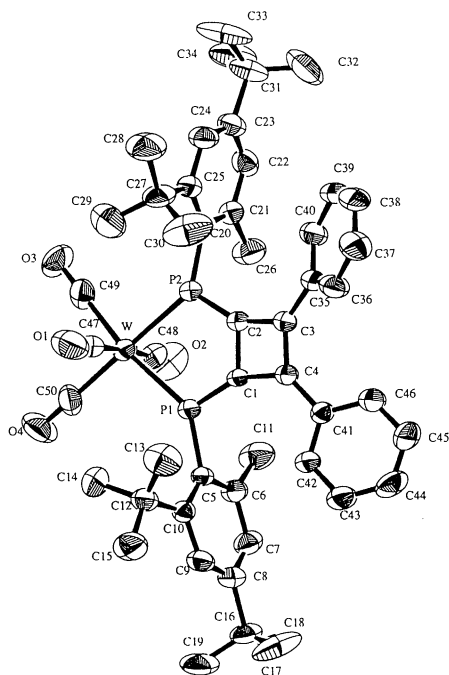


Figure 1. Molecular structure of tungsten complex *syn-5C*. Benzene as a solvent was omitted for clarity.

diphosphinidenecyclobutene **4**, as well as the group-6 metal carbonyl complexes **5**, is less suffering from steric congestion compared with the Ar derivatives.

On the other hand, *anti-5C* was analyzed by HPLC using a Daicel chiral column (Chiralcel OD) and a baseline separation was attained.¹⁵ Each separated enantiomer showed the symmetrical CD spectrum as shown in Figure 2.

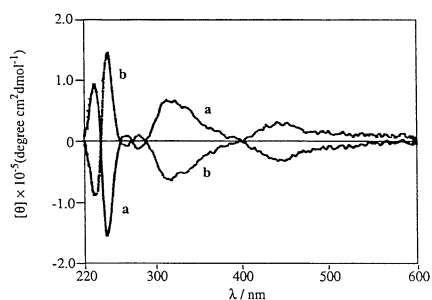


Figure 2. CD spectrum of the separated enantiomers of *anti-5C* (a: first-eluted isomer; b: second-eluted isomer; JASCO J-720, hexane, 1-cm cell).

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References and Notes

- 1 M. Regitz and O. J. Scherer, *Multiple Bonds and Low Coordination in Phosphorus Chemistry*, Georg Thieme Verlag, Stuttgart (1990).
- 2 a) M. Yoshifuji, I. Shima, N. Inamoto, K. Hirotsu, and T. Higuchi, *J. Am. Chem. Soc.*, **103**, 4587 (1981); *J. Am. Chem. Soc.*, **104**, 6167 (1982); b) M. Yoshifuji, K. Toyota, and N. Inamoto, *Tetrahedron Lett.*, **26**, 1727 (1985); c) M. Yoshifuji, K. Toyota, K. Shibayama, and N. Inamoto, *Tetrahedron Lett.*, **25**, 1809 (1984); d) M. Yoshifuji, K. Toyota, and N. Inamoto, *J. Chem. Soc., Chem. Commun.*, **1984**, 689.
- 3 R. Appel, V. Winkhaus, and F. Knoch, *Chem. Ber.*, **120**, 243 (1987).
- 4 G. Märkl, P. Kreitmeier, H. Nöth, and K. Polborn, *Angew. Chem., Int. Ed. Engl.*, **29**, 927 (1990).
- 5 a) M. Yoshifuji, K. Toyota, M. Murayama, H. Yoshimura, A. Okamoto, K. Hirotsu, and S. Nagase, *Chem. Lett.*, **1990**, 2195; b) K. Toyota, K. Tashiro, M. Yoshifuji, and S. Nagase, *Bull. Chem. Soc. Jpn.*, **65**, 2297 (1992); c) K. Toyota, K. Tashiro, T. Abe, and M. Yoshifuji, *Heteroatom Chem.*, **5**, 549 (1994).
- 6 a) K. Toyota, K. Tashiro, and M. Yoshifuji, *Chem. Lett.*, **1991**, 2079; b) K. Toyota, K. Tashiro, M. Yoshifuji, I. Miyahara, A. Hayashi, and K. Hirotsu, *J. Organomet. Chem.*, **431**, C35 (1992).
- 7 K. Toyota, K. Masaki, T. Abe, and M. Yoshifuji, *Chem. Lett.*, **1995**, 221.
- 8 M. Yoshifuji, K. Shibayama, N. Inamoto, T. Matsushita, and K. Nishimoto, *J. Am. Chem. Soc.*, **105**, 2495 (1983).
- 9 M. Yoshifuji, S. Ito, K. Toyota, and M. Yasunami, *Heteroatom Chem.*, **7**, 23 (1996).
- 10 M. Baudler and J. Simon, *Chem. Ber.*, **121**, 281 (1988).
- 11 G. Yu. Mikhailov, I. G. Trostyanskaya, M. A. Kazankova, and I. F. Lutsenko, *Zh. Obshch. Khim.*, **57**, 2636 (1987).
- 12 *anti-5A*: $\delta p = 189.5$; *syn-5A*: $\delta p = 187.5$. *anti-5B*: $\delta p = 169.4$; *syn-5B*: $\delta p = 168.0$. *anti-5C*: dark purple solid, $\delta p = 146.5$ ($J_{PW} = 254$ Hz); *syn-5C*: dark purple, mp 212 °C (decomp.), $\delta p = 145.1$ ($J_{PW} = 254$ Hz); *anti-5C+syn-5C*: 37% yield based on **1** (0.62 g, 3.1 mmol, employed). The peaks due to the carbonyl carbon in ¹³C NMR were diagnostic of the structure of the complexes and the major one was assigned to the *anti*-type which has two apical CO's and two equatorial CO's, whereas the minor one was assigned to the *syn*-type, which has two equatorial CO's and two apical non-equivalent CO's.
- 13 Crystal Data for *syn-5C*: Recrystallized from benzene-hexane. C₅₀H₅₆O₄P₂W·1/2C₆H₆, Mr = 1005.84, prismatic, $a = 11.485(3)$ Å, $b = 21.434(3)$ Å, $c = 11.270(2)$ Å, $\alpha = 95.83(2)^\circ$, $\beta = 112.85(2)^\circ$, $\gamma = 97.02(2)^\circ$, $V = 2503(1)$ Å³, P1̄ (#2), $Z = 2$, $T = 296$ K, $R = 0.045$, $R_w = 0.043$; 7430 unique reflections with $I > 3\sigma(I)$. The structure was solved with SHELXS86; G. M. Sheldrick, *SHELXS86, Programs for the Automatic Solution of Crystal Structures*, University of Göttingen, Germany (1986). Further details of the crystal structure investigation for *syn-5C* are available on request from the Director of the Cambridge Crystallographic Data Centre, 12 Union Road, GB-Cambridge CB2 1EZ (UK).
- 14 Some selected bond lengths (Å) and angles (°) of *syn-5C*: W–P(1), 2.515(2); W–P(2), 2.489(2); W–C(47), 1.955(9); W–C(48), 1.930(9); W–C(49), 1.986(8); W–C(50), 2.014(9); P(1)–C(1), 1.669(6); P(1)–C(5), 1.833(6); P(2)–C(2), 1.662(6); P(2)–C(20), 1.821(7); C(1)–C(2), 1.507(8); C(1)–C(4), 1.483(8); C(2)–C(3), 1.466(8); C(3)–C(4), 1.410(9); P(1)–W–P(2), 77.73(6); P(1)–W–C(47), 90.7(2); P(1)–W–C(48), 91.0(2); P(1)–W–C(49), 169.9(2); P(1)–W–C(50), 99.4(2); P(2)–W–C(47), 90.2(2); P(2)–W–C(48), 94.5(3); P(2)–W–C(49), 92.2(2); P(2)–W–C(50), 173.5(3); C(47)–W–C(48), 175.2(4); C(47)–W–C(49), 89.9(3); C(47)–W–C(50), 83.9(4); C(48)–W–C(49), 89.2(3); C(48)–W–C(50), 91.4(4); C(49)–W–C(50), 90.6(3); W–P(1)–C(1), 111.5(2); W–P(1)–C(5), 137.7(2); C(1)–P(1)–C(5), 110.1(3); W–P(2)–C(2), 112.1(2); W–P(2)–C(20), 135.6(2); C(2)–P(2)–C(20), 111.8(3); P(1)–C(1)–C(2), 118.9(4); P(1)–C(1)–C(4), 153.2(5); C(2)–C(1)–C(4), 87.8(5); P(2)–C(2)–C(1), 119.7(5); P(2)–C(2)–C(3), 151.9(5); C(1)–C(2)–C(3), 88.4(5); C(2)–C(3)–C(4), 92.3(5); C(2)–C(3)–C(35), 130.0(6); C(4)–C(3)–C(35), 137.2(6); C(1)–C(4)–C(3), 91.5(5); C(1)–C(4)–C(41), 132.8(6); C(3)–C(4)–C(41), 135.7(6); P(1)–C(5)–C(6), 113.4(5); P(1)–C(5)–C(10), 126.2(5).
- 15 Column: Chiralcel OD 25 × 0.46 (i.d.) cm (Daicel); eluent: hexane; flow speed: 0.5 ml/min; temperature: RT. *anti-5C* was dissolved in a mixture of hexane-isopropyl alcohol (9:1) in 0.1 mg/mL. First peak appeared at 10.4 min and the second one at 11.6 min.